oxides and oxyacids of sulphur, thiosulphuric acid and salts, thionic acids and their salts, tetrasulphur tetranitride.

**Group-17** - General properties, hydrogen halides, oxides and oxyacids of halogens, interhalogen compounds, polyhalides, basic properties of halogens.

#### UNIT - V

**Non -Aqueous Solvents -** Physical properties of a solvent, types of solvents and their general characteristics, Differentiating and leveling solvents, reactions in non-aqueous solvents with special reference to liquid  $NH_3$  and liquid  $SO_2$ .

Acids and Bases - Arrhenius, Bronsted - Lowry, Lux - Flood, solvent system and Lewis concepts of acid and bases, Usanovitch definition.

## Books Recommended

- 1. Concise Inorganic Chemistry: J. D. Lee
- 2. General Inorganic Chemistry: J. A. Duffy, Longman (2nd Ed.)
- 3. Principles of Inorganic Chemistry: B. R. Puri and L. R. Sharma
- 4. Basic Inorganic Chemistry: F.A.Cotton and G. Wilkinson, Wiley Eastern
- 5. Molecular Geometry: R. J. Gillespie, Van Nostrand Reinhold.
- 6. Inorganic Chemistry (Hindi ed.): Suresh Ameta, A. Sharma and M. Mehta, Himanshu Pub.

# PAPER II ORGANIC CHEMISTRY

Time-3 Hrs.

M.M. 50

NOTE: The paper will be divided into THREE sections.

Section-A Ten questions (short type answer) two from each Unit will be asked. Each question will be of half mark and the candidates are required to attempt all questions.

Total 5 marks

Section-B Five questions (answer not exceeding 250 words) one from each Unit with internal choice will be asked and the candidates are required to attempt all questions. Each question will be of 5 marks.

Total 25 marks

Section-C Four questions may be in parts covering all the five Units (answer not exceeding 500 words) will be asked. The candidates are required to attempt any **TWO** questions. Each question will be of 10 marks.

Total 20 marks

### UNIT-I

Structure and Bonding - Localized and delocalized chemical bond, Van der Waals interaction, charge transfer complexes, resonance, hyperconjugation, aromaticity, electromeric, inductive and field effects, hydrogen bonding.

Mechanism of Organic Reactions - Curved arrow Conformational isomerism- conformational analysis of notation, drawing electron movements with arrows, ethane and n-butane; conformations of cyclohexane, half-headed and double- headed arrows, types of axial and equitorial bonds, conformation of mono organic reactions, energy considerations.

Reactive Intermediates - Carbocations, carbanions, Wedge formulae. free radicals, carbenes, arynes and nitrenes, their Difference between configuration and conformation. formation and stabilities.

Methods of determination of reaction mechanism (product analysis, intermediates, isotope effects, kinetic and stereochemical studies).

#### UNIT-II

Stereochemistry of Organic Compounds of isomerism, types of isomerism.

Optical Isomerism- Elements of symmetry, molecular of cyclopropane ring: banana bond. chirality, enantiomers, stereogenic centre, optical activity, properties of enantiomers, chiral and achiral stereogenic two retention and racemization.

Relative and absolute configuration, sequence rules D and L, R and S systems of nomenclature.

substituted cyclohexane derivatives. Newman projection and Sawhorse formulae, Fischer and Flying

#### UNIT-III

Alkanes: General methods of formation, physical & chemical properties. Mechanism of free radical substitution in alkanes with reference to halogenation, orientation, reactivity and selectivity.

Cycloalkanes - Nomenclature, methods of formation, - Concept chemical reactions, Baeyer's strain theory and its limitation, ring strain in small rings (cyclopropane and cyclobutane), theory of strainless rings, the case

Alkenes, Dienes and Alkynes - Brief introduction of alkenes, their formation with reference to centers mechanisms of dehydration of alcohols and diastereomers, three and erythre diastereomers, meso dehydrohalogenation of alkyl halides, regioselectivity compounds, resolution of enantiomers, inversion in alcohol dehydration The Saytzeff rule, Hofmann elimination, physical properties and relative stabilities of alkenes.

Chemical reactions of alkenes- mechanisms involved in hydrogenation, electrophilic and free radical Geometric isomerism- determination of configuration additions, Markownikoff's rule, hydroborationof geometric isomers. E and Z system of nomenclature oxidation, oxymercuration-reduction, epoxidation, geometric isomerism in oximes and alicyclic compounds ozonolysis, hydration, hydroxylation and oxidation with KMnO, polymerization of alkenes, substitution at

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UNIT -V

applications of ethylene and propene.

Nomenclature and classification of dienes: isolate alkyl and Aryl Halides - Nomenclacture and classes Alder reaction.

Alkynes: Acidity of alkynes, mechanism electrophilic and nucleophilic addition reactions Haloform reaction, Freons: hydroboration, metal - ammonia reductions, oxidatio and polymerization.

#### Unit - IV

Arenes and Aromaticity - Nomenclature of benzen derivatives, the aryl group, aromatic nucleus and sid Relative reactivites of alkyl halides v/s allyl, vinyl chain, structure of benzene, molecular formula an and aryl halides. synthesis and uses of DDT and BHC. Kekule structure, stability and carbon - carbon bon lengths of benzene, resonance structure and M. Books Recommended

Aromaticity: The Huckel rule, aromatic ions. Aromatic electrophilic substitution: General pattern of the mechanism, role of s and p complexes. Mechanism 2. Modern Principles of Organic Chemistry: M. K. Jain nitration, halogenation, sulphonation, mercuration and Friedel -Craft reaction. Energy profile diagrams Activating and deactivating substituents, orientations. and ortho-para ratio. Side chain reactions of benzen derivatives. Birch reduction, Methods of formation and chemical reactions of alkylbenzenes, alkynylbenzen 4. A Text Book of Organic Chemistry: B. S. Bahl and

conjugated and cumulated dienes, structure of allenes alkyl halides, methods of formation, chemical and butadiene, methods of formation, polymerization eactions, mechanism of nucleophilic substitution chemical reactions - 1,2 and 1,4 - additions, Diels reactions of alkyl halides,  $S_N^2$  and  $S_N^1$  reactions with energy profile diagrams, factors affecting  $S_N^2$  and  $S_N^1$ reactions.

Methods of formation of aryl halides, nuclear and side chain reactions, the addition - elimination and elimination - addition reaction, mechanisms of mucleophilic aromatic substitution reactions.

- 1. A Text Book of Organic Chemistry: K. S. Tiwari, S. N. Mehrotra and N. K. Vishnoi.
- and S.C. Sharma
- A Text Book of Organic Chemistry: (Vol. I & II) O. P. Agarwal,
- Arun Bahl.